# 2,3,10,11-Tetraphenyl-1,4,9,12-tetraoxadi-spiro[4.2.4.2]tetradeca-6,13-diene 

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(Received 30 January 1995; accepted 4 September 1995)


#### Abstract

The crystal structure of the title compound, $\mathrm{C}_{34} \mathrm{H}_{28} \mathrm{O}_{4}$, consists of discrete molecules held together by normal van der Waals interactions. The molecule possesses a crystallographic twofold axis passing through the midpoints of the two double bonds of the cyclohexadiene ring. The asymmetric C atoms in the dioxolane rings have an $S, S$ configuration.


## Comment

The title compound, (1), was obtained as a minor product ( $14 \%$ yield, m.p. $475-477 \mathrm{~K}$ ) from the reaction of 4,4-dimethoxycyclohexadienone ( 0.015 mol ) and $(S, S)$-hydrobenzoin ( 0.0163 mol ) in the presence of $\mathrm{Et}_{2} \mathrm{O} . \mathrm{BF}_{3}(0.031 \mathrm{~mol})$ in glyme at 258 K . The product was separated from the major component, 1,4-dioxa-2,3-diphenylspiro[5.4]deca-6,9-dien-8-one ( $32 \%$ yield, m.p. 457 K ), and purified by column chromatography (neutral silica) using gradient elution starting with $100 \%$ petroleum spirit ( $313-333 \mathrm{~K}$ ) to $100 \%$ dichloromethane. Both compounds were characterized by IR, ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectroscopy and mass spectral analysis.

(1)

The structure of (1) was confirmed by an X-ray study. The absolute configuration could not be determined from this study because of the absence of significant anomalous scattering, but it was implied from the known chirality of the materials used in the synthesis [Flack (1983) parameter $=-2.3(13)]$. The molecular structure
of (1) is shown in Fig. 1, which also indicates the atom labelling in the asymmetric unit. The molecule possesses a crystallographic twofold axis passing through the midpoints of the $\mathrm{C}(16)-\mathrm{C}\left(16^{\prime}\right)$ and $\mathrm{C}(17)-\mathrm{C}\left(17^{\prime}\right)$ bonds of the central cyclohexadiene ring. This ring and the terminal phenyl groups are planar within 0.013 (3) $\AA$ and have normal geometric parameters, but the dioxolane ring, as expected, is non-planar. The orientations of the phenyl rings relative to their particular OCC moieties of the dioxolane ring are somewhat different, as indicated by the torsion angles $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(3)-\mathrm{C}(4) 31.2(2)$ and $\mathrm{O}(2)-\mathrm{C}(2)-$ $\mathrm{C}(9)-\mathrm{C}(10) 42.0(2)^{\circ}$.


Fig. 1. The molecular structure of (1) showing the atom labelling in the asymmetric unit. Ellipsoids are plotted at the $50 \%$ probability level.

It is pertinent to note that the cis configuration and $C_{2}$ symmetry of this molecule, which is of course a direct consequence of the incorporation of the two ( $S, S$ )hydrobenzoins, contrasts with the trans configuration of the unsubstituted centrosymmetric achiral compound 1,4,9,12-tetraoxadispiro[4.2.4.2]tetradecane, (2) (Chadwick, Dunitz \& Schweizer, 1977]. These differences are highlighted in Fig. 2. The conformation of the fivemembered dioxolane ring is best described as a slightly twisted envelope (Altona, Geize \& Romers, 1968) and resembles that in compound (2). The torsion angles in the ring are $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{O}(2) 29.6(2), \mathrm{C}(1)-$ $\mathrm{C}(2)-\mathrm{O}(2)-\mathrm{C}(15)-38.1(2), \mathrm{C}(2)-\mathrm{O}(2)-\mathrm{C}(15)-$ $\mathrm{O}(1) 31.8(2), \mathrm{C}(1)-\mathrm{O}(1)-\mathrm{C}(15)-\mathrm{O}(2)-11.8(2)$ and
$\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{O}(1)-\mathrm{C}(15)-11.2(2)^{\circ}$; the corresponding values in the unsubstituted saturated compound (2) are $27.7,-31.9,23.9,-5.8$ and $-13.4^{\circ}$, respectively. The two sets of values show a common trend and the differences $\left(2-8^{\circ}\right)$ may be attributed to the presence of the bulky phenyl substituents in compound (1). Other structural parameters are as expected for this type of compound. The intermolecular distances are appropriate for van der Waals interactions.


Fig. 2. The structures of compounds (1) and (2) showing the cis and trans conformations of the $\mathrm{C}-\mathrm{C}$ ends of the dioxolane rings.

## Experimental

## Crystal data

$\mathrm{C}_{34} \mathrm{H}_{28} \mathrm{O}_{4}$
$M_{r}=500.56$
Trigonal
$P 3_{2} 21$
$a=9.691$ (2) $\AA$
$c=24.841(6) \AA$
$V=2020.2(8) \AA^{3}$
$Z=3$
$D_{x}=1.234 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ not measured

## Data collection

Delft Instruments FAST area-detector diffractometer
Measurement method: Darr, Drake, Hursthouse \& Malik (1993)
Absorption correction: none

## Refinement

Refinement on $F^{2}$
$R(F)=0.0309$
$w R\left(F^{2}\right)=0.0629$
$S=0.892$
2164 reflections
228 parameters
H atoms refined isotropically
$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0195 P)^{2}\right]$

Atomic scattering factors from International Tables for Crystallography (1992, Vol. C, Tables 4.2.6.8 and 6.1.1.4)
where $P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3$
Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters $\left(\AA^{2}\right)$

| $U_{\mathrm{eq}}=(1 / 3) \sum_{i} \sum_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$ |  |  |  |  |
| :--- | :---: | :---: | :---: | :---: |
| $\quad \boldsymbol{y}$ | $z$ | $U_{\mathrm{eq}}$ |  |  |
| $\mathrm{O}(1)$ | $0.38457(15)$ | $0.89924(14)$ | $0.06650(5)$ | $0.0335(3)$ |
| $\mathrm{O}(2)$ | $0.28960(15)$ | $0.70660(13)$ | $0.13176(4)$ | $0.0324(3)$ |
| $\mathrm{C}(1)$ | $0.3362(2)$ | $0.7499(2)$ | $0.03983(7)$ | $0.0280(4)$ |
| $\mathrm{C}(2)$ | $0.3295(2)$ | $0.6406(2)$ | $0.08617(7)$ | $0.0285(4)$ |
| $\mathrm{C}(3)$ | $0.4498(2)$ | $0.7706(2)$ | $-0.00524(7)$ | $0.0267(4)$ |
| $\mathrm{C}(4)$ | $0.6105(2)$ | $0.8854(2)$ | $-0.00196(8)$ | $0.0325(5)$ |
| $\mathrm{C}(5)$ | $0.7148(2)$ | $0.9055(2)$ | $-0.04297(8)$ | $0.0396(5)$ |
| $\mathrm{C}(6)$ | $0.6608(3)$ | $0.8082(3)$ | $-0.08825(8)$ | $0.0405(5)$ |
| $\mathrm{C}(7)$ | $0.5018(3)$ | $0.6924(3)$ | $-0.09192(8)$ | $0.0377(5)$ |
| $\mathrm{C}(8)$ | $0.3972(2)$ | $0.6740(2)$ | $-0.05079(7)$ | $0.0321(5)$ |
| $\mathrm{C}(9)$ | $0.2092(2)$ | $0.4661(2)$ | $0.08189(7)$ | $0.0276(5)$ |
| $\mathrm{C}(10)$ | $0.0483(2)$ | $0.4107(2)$ | $0.08956(8)$ | $0.0332(5)$ |
| $\mathrm{C}(11)$ | $-0.0606(3)$ | $0.2491(2)$ | $0.08699(8)$ | $0.0377(5)$ |
| $\mathrm{C}(12)$ | $-0.0108(3)$ | $0.1419(3)$ | $0.07689(8)$ | $0.0409(5)$ |
| $\mathrm{C}(13)$ | $0.1492(3)$ | $0.1962(2)$ | $0.06840(9)$ | $0.0469(6)$ |
| $\mathrm{C}(14)$ | $0.2582(3)$ | $0.3574(2)$ | $0.07111(8)$ | $0.0405(5)$ |
| $\mathrm{C}(15)$ | $0.3748(2)$ | $0.8750(2)$ | $0.12464(7)$ | $0.0301(5)$ |
| $\mathrm{C}(16)$ | $0.2799(2)$ | $0.9429(2)$ | $0.14825(7)$ | $0.0323(5)$ |
| $\mathrm{C}(17)$ | $0.5389(2)$ | $0.9448(2)$ | $0.14762(7)$ | $0.0308(4)$ |

Table 2. Selected geometric parameters $\left(\AA^{\circ},^{\circ}\right)$

| $\mathrm{O}(1)-\mathrm{C}(1)$ | $1.440(2)$ | $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.386(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O}(1)-\mathrm{C}(15)$ | $1.459(2)$ | $\mathrm{C}(9)-\mathrm{C}(14)$ | $1.382(3)$ |
| $\mathrm{O}(2)-\mathrm{C}(15)$ | $1.424(2)$ | $\mathrm{C}(9)-\mathrm{C}(10)$ | $1.385(3)$ |
| $\mathrm{O}(2)-\mathrm{C}(2)$ | $1.445(2)$ | $\mathrm{C}(10)-\mathrm{C}(11)$ | $1.385(3)$ |
| $\mathrm{C}(1)-\mathrm{C}(3)$ | $1.512(2)$ | $\mathrm{C}(11)-\mathrm{C}(12)$ | $1.370(3)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.544(2)$ | $\mathrm{C}(12)-\mathrm{C}(13)$ | $1.382(3)$ |
| $\mathrm{C}(2)-\mathrm{C}(9)$ | $1.502(2)$ | $\mathrm{C}(13)-\mathrm{C}(14)$ | $1.382(3)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.392(3)$ | $\mathrm{C}(15)-\mathrm{C}(16)$ | $1.493(3)$ |
| $\mathrm{C}(3)-\mathrm{C}(8)$ | $1.392(2)$ | $\mathrm{C}(15)-\mathrm{C}(17)$ | $1.495(3)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.379(3)$ | $\mathrm{C}(16)-\mathrm{C}\left(16^{1}\right)$ | $1.325(4)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.391(3)$ | $\mathrm{C}(17)-\mathrm{C}\left(17^{1}\right)$ | $1.325(3)$ |
| $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.383(3)$ |  |  |
| $\mathrm{C}(1)-\mathrm{O}(1)-\mathrm{C}(15)$ | $109.34(12)$ | $\mathrm{C}(14)-\mathrm{C}(9)-\mathrm{C}(10)$ | $118.7(2)$ |
| $\mathrm{C}(15)-\mathrm{O}(2)-\mathrm{C}(2)$ | $106.23(13)$ | $\mathrm{C}(14)-\mathrm{C}(9)-\mathrm{C}(2)$ | $120.2(2)$ |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(3)$ | $110.98(14)$ | $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(2)$ | $121.1(2)$ |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | $103.14(13)$ | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $120.4(2)$ |
| $\mathrm{C}(3)-\mathrm{C}(1)-\mathrm{C}(2)$ | $114.90(15)$ | $\mathrm{C}(12)-\mathrm{C}(11)-\mathrm{C}(10)$ | $120.5(2)$ |
| $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{C}(9)$ | $108.96(14)$ | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)$ | $119.6(2)$ |
| $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{C}(1)$ | $102.09(13)$ | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | $120.1(2)$ |
| $\mathrm{C}(9)-\mathrm{C}(2)-\mathrm{C}(1)$ | $117.53(15)$ | $\mathrm{C}(9)-\mathrm{C}(14)-\mathrm{C}(13)$ | $120.7(2)$ |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(8)$ | $118.2(2)$ | $\mathrm{O}(2)-\mathrm{C}(15)-\mathrm{O}(1)$ | $105.15(13)$ |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(1)$ | $120.8(2)$ | $\mathrm{O}(2)-\mathrm{C}(15)-\mathrm{C}(16)$ | $108.91(15)$ |
| $\mathrm{C}(8)-\mathrm{C}(3)-\mathrm{C}(1)$ | $121.0(2)$ | $\mathrm{O}(1)-\mathrm{C}(15)-\mathrm{C}(16)$ | $108.80(15)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(3)$ | $121.2(2)$ | $\mathrm{O}(2)-\mathrm{C}(15)-\mathrm{C}(17)$ | $110.33(14)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $120.1(2)$ | $\mathrm{O}(1)-\mathrm{C}(15)-\mathrm{C}(17)$ | $109.67(14)$ |
| $\mathrm{C}(7)-\mathrm{C}(6)-\mathrm{C}(5)$ | $119.5(2)$ | $\mathrm{C}(16)-\mathrm{C}(15)-\mathrm{C}(17)$ | $113.6(2)$ |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | $120.2(2)$ | $\mathrm{C}(16)-\mathrm{C}(16)-\mathrm{C}(15)$ | $123.20(11)$ |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(3)$ | $120.9(2)$ | $\mathrm{C}(17)-\mathrm{C}(17)-\mathrm{C}(15)$ | $123.15(10)$ |
| $\mathrm{S}(1)$ |  |  |  |

Symmetry code: (i) $1+x-y, 2-y, \frac{1}{3}-z$.
The unit-cell and intensity data were recorded at 140 K on a FAST area-detector diffractometer using the routines ENDEX, REFINE and MADONL in the MADNES software (Pflugrath \& Messerschmidt, 1989); detailed procedures are described by Darr, Drake, Hursthouse \& Malik (1993). The H atoms were all located from a difference map and refined isotropically. All calculations were performed on a 486DX2/66 personal computer.

Data reduction: $A B S M A D$ (Karaulov, 1992). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: SNOOPI (Davies, 1983). Software used to prepare material for publication: SHELXL93.

The authors thank the EPSRC for support of the Xray Crystallography Service at UWC.

Lists of structure factors, anisotropic displacement parameters, H atom coordinates and complete geometry have been deposited with the IUCr (Reference: SZ1047). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Acta Cryst. (1996). C52, 381-383

# 2,3,4,5-Tetra- $O$-acetyl-1,6-di- $O$-(triphenyl-methyl)-D-mannitol 

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(Received 19 September 1994; accepted 4 October 1995)


#### Abstract

The central six-C-atom chain of the title compound, $\mathrm{C}_{52} \mathrm{H}_{50} \mathrm{O}_{10}$, adopts a nearly planar extended conformation free from $\mathrm{C} / / \mathrm{C}, \mathrm{C} / / \mathrm{O}$ and $\mathrm{O} / / \mathrm{O}$ 1,3-parallel interactions. The three torsion angles formed by these atoms have values of $177.4(2), 174.2(2)$ and $-178.0(2)^{\circ}$. The bulky triphenylmethyl groups, which are oriented gauche to their neighboring acetoxy groups with O -


$\mathrm{C}-\mathrm{C}-\mathrm{O}$ torsion angles of -69.8 (2) and $-63.0(2)^{\circ}$, cause no distortion of the bond lengths and bond angles of the sugar moiety.

## Comment

The title compound, 2,3,4,5-tetra- $O$-acetyl-1,6-di- $O$-(tri-phenylmethyl)-D-mannitol, (I), is one of the acyclic alditol derivatives which have have shown, by crystal structure analyses, to possess some interesting conformational features (Kopf, Morf, Zimmer, Bischoff \& Koll 1992). The structure of (I) was determined in order to confirm the identity of the reaction product and to observe to what extent the conformation of its parent compound, D-mannitol, (II) (Kim, Jeffrey \& Rosenstein 1968; Kanters, Roelofsen \& Smiths 1977), might be affected by the presence of the two bulky triphenylmethyl groups.

(I)

(II)

$$
R=\text { triphenylmethyl }
$$

The molecular structure of (I) (Fig. 1) reveals that the molecule does not possess crystallographic twofold symmetry as observed in the parent compound (II). Approximate twofold symmetry is present, however, and the mannitol core has an extended conformation with no 1,3-parallel interactions between the heavy atoms. This conformation is described by the three anti torsion angles given in the Abstract, as well as $\mathrm{O} 1-\mathrm{C} 1-$ $\mathrm{C} 2-\mathrm{C} 3$ and C4-C5-C6-O6 [47.7 (2) and 53.9 (2) ${ }^{\circ}$, respectively]. The $\mathrm{C}-\mathrm{C}$ bond distances in the sugar moiety are in the range 1.514 (3)-1.526(3) $\AA$ while the $\mathrm{C}-\mathrm{O}$ bonds are in the range 1.411 (2) -1.445 (2) $\AA$, in good agreement with the corresponding values for (II) and other related carbohydrate structures (Jeffrey \& French, 1978; Allen, 1986; Allen, Kennard, Watson, Brammer, Orpen \& Taylor, 1987). The orientation of the triphenylmethyl groups at the terminal positions ( C 1 and C6) are gauche to their neighbouring acetoxy O atoms, as are the primary hydroxy groups in the parent compound (II), the torsion angles $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{O} 2$ and $\mathrm{O} 6-\mathrm{C} 6-\mathrm{C} 5-\mathrm{O} 5$ being $-69.8(2)$ and $-63.0(2)^{\circ}$, respectively. The four acetoxy groups, on $\mathrm{C} 2, \mathrm{C} 3, \mathrm{C} 4$, and C5, are oriented in the sequence trans, gauche, trans; the torsion angles $\mathrm{O} 2-\mathrm{C} 2-\mathrm{C} 3-\mathrm{O} 3, \mathrm{O} 3-\mathrm{C} 3-$ $\mathrm{C} 4-\mathrm{O} 4$ and $\mathrm{O} 5-\mathrm{C} 5-\mathrm{C} 4-\mathrm{O} 4$ are 178.5 (2), 49.9 (2)

